

## VPython Class 9: Fitting wavefunctions into “boxes” with a computer

### 1. Introduction

In class last week we talked about using Schrödinger’s equation to find special-case quantum wave functions with definite energies. Today we introduce technique using a computer, called numerical integration, to *construct* such special-case wavefunctions point by point.

Recall that Schrödinger’s equation is

$$-\frac{\hbar^2}{2m} \frac{d^2\psi(x)}{dx^2} + U(x)\psi(x) = E\psi(x). \quad (1)$$

I’m now going to use a standard trick to make things easier for computer work: I’m going to pick a set of dimensionless variables. In the first part of the lab we’re going to look at the one-dimensional “particle in a box” problem, or square-well potential, with the well having a length  $L$ , but I don’t want to worry about whether  $L = 1$  nm, or  $L = 0.3$  nm, or something else, so I choose a new variable

$$X \equiv \frac{x}{L}. \quad (2)$$

The left end of the “box” will then be at  $X = 0$ , and the right end will be at  $X = 1$ . This choice also affects the expression for the second derivative. Since  $x = XL$ , we have

$$\frac{d^2\psi}{dx^2} = \frac{1}{L^2} \frac{d^2\psi}{dX^2}. \quad (3)$$

Using this, and some re-arrangement, give Schrödinger’s equation as

$$-\frac{d^2\psi(X)}{dX^2} + \frac{2mL^2}{\hbar^2} U(X)\psi(X) = \frac{2mL^2}{\hbar^2} E\psi(X). \quad (4)$$

or

$$-\frac{d^2\psi(X)}{dX^2} + \frac{8mL^2\pi^2}{h^2} U(X)\psi(X) = \frac{8mL^2\pi^2}{h^2} E\psi(X). \quad (5)$$

This suggests defining new dimensionless variables for the energies  $E$  and  $U$  in terms of the energy  $h^2/8mL^2$ :

$$E' = \frac{E}{\frac{h^2}{8mL^2}} = E \frac{8mL^2}{h^2} \quad \text{and} \quad U' = \frac{U}{\frac{h^2}{8mL^2}} = U \frac{8mL^2}{h^2}. \quad (6)$$

Using these expressions, Schrödinger’s equation becomes

$$-\frac{d^2\psi(X)}{dX^2} + \pi^2 U'(X) \psi(X) = \pi^2 E' \psi(X). \quad (7)$$

NOTE: I could just as easily have incorporated the factor of  $\pi^2$  within my definition of  $E'$  and  $U'$ , but by not doing so some things come out a little cleaner in this exercise.

This is now in a form that a computer can evaluate, without the messy details of any constants hanging around. Of course at the end, all dimensionless quantities like  $E'$  will have to be converted back to energies with units. That’s easy though: just multiply  $E'$  by  $\frac{h^2}{8mL^2}$ .

## 2. Solving Schrödinger's equation numerically

Rearrangement of Eq. (7) highlights the fact that Schrödinger's equation gives us information about the second derivative of the wavefunction:

$$\frac{d^2\psi(x)}{dx^2} = -\pi^2 [E - U(x)] \psi(x), \quad (8)$$

where I have switched back to conventional symbols  $x$ ,  $U$  and  $E$ , but with the understanding that they represent the dimensionless variables  $X$ ,  $E'$ , and  $U'$ .

For a computer to build a solution to this equation we need a numerical approximation to the second derivative that we can use at a set of discrete points. We will use an equally spaced set of value of  $x$  given by

$$x_i = x_0 + i\Delta, \quad (9)$$

where  $i$  is an integer. A set of such  $x$  values can be created using the `linspace` function from `numpy`. The corresponding values of the function will be labeled

$$y_i = f(x_i). \quad (10)$$

You are already familiar with a numerical approximation to the first derivative:

$$\left. \frac{dy}{dx} \right|_{x_i} \simeq \frac{y_{i+1} - y_i}{\Delta}. \quad (11)$$

For the second derivative we can use this approximation for the first derivative in the intervals to the right and left of  $x_i$ :

$$\begin{aligned} \left. \frac{d^2y}{dx^2} \right|_{x_i} &\simeq \frac{\left. \frac{dy}{dx} \right|_{\text{right}} - \left. \frac{dy}{dx} \right|_{\text{left}}}{\Delta} \\ &\simeq \frac{\frac{y_{i+1} - y_i}{\Delta} - \frac{y_i - y_{i-1}}{\Delta}}{\Delta} \\ &= \frac{y_{i+1} - 2y_i + y_{i-1}}{\Delta^2} \end{aligned} \quad (12)$$

Solving this for  $y_{i+1}$  gives

$$y_{i+1} = 2y_i - y_{i-1} + \Delta^2 \left. \frac{d^2y}{dx^2} \right|_{x_i}. \quad (13)$$

If we know  $y_0$  and  $y_1$ , and we know the second derivative from Schrödinger's equation, we can calculate  $y_2$ . Then we (i.e., the computer) can repeat the process over and over to calculate  $y_3$ ,  $y_4$ , etc.

### 2.1 Particle in a box

- Locate a graphing program you have written, and make sure that it still works. In the following I'm assuming that you have have a Python function, and that it has a definition beginning with a statement like

```
def function_name:
    ...
```

- Add a potential energy function at the top of your program. This is ridiculously simple for the first part of this exercise when we're considering a particle in a box:

```
def u(x):
    return 0
```

- Make the left endpoint for your graph be  $x = 0$ , and your right endpoint be  $x = 1$ .
- Choose to graph at least 101 points.
- Create the array of  $x$  values using

```
x = np.linspace(lep, rep, n)
```

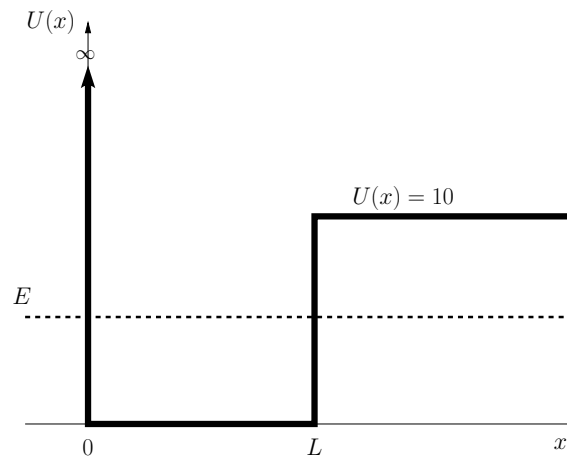
where I have used the variables `lep`, `rep`, and `n` for my left endpoint, right endpoint, and number of points respectively. (Remember that for  $n$  points, there are  $n - 1$  intervals.)

- Calculate the distance between your points and define this to be `dx`.
- Modify your function so that it includes a “guess” for the dimensionless energy, the declaration of an array for your  $y$  values (initially all set to zero), an initial value for `y[1]`, and a `for` loop to calculate the rest of the  $y$  values:

```
def integrate(x):
    '''Function to perform numerical integration'''
    e = 0.8 # Value of dimensionless energy
    y = np.zeros(len(x)) # Create array for y values
    y[1] = dx # Initialize y[1]
    for i in range(1,np-1): # Calculate y[2], y[3], y[4], etc.
        YOUR CODE HERE
    return y
```

- At this point you should have the plot of a wavefunction corresponding to your first energy “guess” `e = 0.8`. Is this a valid wavefunction for a particle in a box? If not, what do you have to adjust to make it a valid wavefunction.
- Use your program to find the three lowest energies of the particle in a box. Convert the values you get to quantities with dimensions. Is this what you expect?

## 2.2 Particle in a semi-infinite square well



- Modify your potential energy function to correspond to that of a semi-infinite square well potential:

```
def u(x):  
    if x<1:  
        return 0  
    else:  
def u(x):  
    if x<1:  
        return 0  
    else:  
        return 10
```

- Make the right endpoint for your graphs be three, and increase the number of points accordingly.
- Adjust the  $y$  limits for your graph to be  $\pm 2$ .
- Do your previous values of the energy work?
- Find the lowest two allowed energies in this potential.