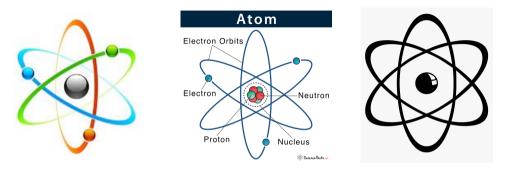


COME AND EXPERIENCE FUN EXPERIMENTS WITH YOUR FAVORITE PHYSICS & ASTRONOMY FACULTY

### Common picture of an atom:



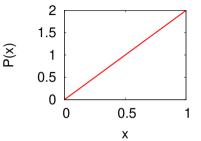
This kind of atom could not exist! The accelerating electron would radiate away all of its energy as an EM wave, and then crash into the nucleus.

#### Classical mechanics + $E\&M \Rightarrow$ atoms can't exist!!

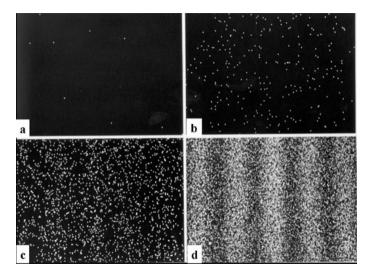
 $\dots$  but fortunately atoms **do** exist. We have to give up on classical mechanics to explain them.

A particle is associated with a probability density P(x) = 2x from x = 0 to x = 1, and P(x) = 0 for all other values of x.

What is the probability that the particle would be found between x = 0 and x = 1/2?

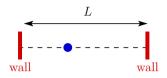


#### **Electron Double Slit Experiment**



https://www.hitachi.com/rd/research/materials/quantum/doubleslit/index.html

You trap a particle in a very small 1D box such that the spread in position  $\sigma_x$  must be quite small. What does Heisenberg's Uncertainty Principle imply about this particle?



Answer each statement with 1 = True or 2 = False.

- A. A small spread in position  $\sigma_x$  implies a small spread in momentum  $\sigma_{p_x}$  (and therefore velocity).
- B. There is a minimum kinetic energy greater than zero that the particle must have.
- C. If we make the box smaller, the kinetic energy of the particle becomes smaller

Given the equation 
$$\frac{df(x)}{dx} + 6x - 2 = 0$$
, test the trial solution  $f(x) = Ax + B$ 

to see if it works. If so, determine the values of the constants  ${\cal A}$  and  ${\cal B}$  needed to make it work.

- **1.** This doesn't work for all values of x.
- **2.** Works if A = 6 and B = -2
- **3.** Works if A = -6 and B = 2.

- 4. Works if A = 2 and B = -6
- **5.** Works if A = -2 and B = 6
- **6.** Works if  $A = \pi$  and  $B = \sqrt{17}$ .

Given the equation 
$$\frac{df(x)}{dx} + 6x - 2 = 0$$
, test the trial solution 
$$f(x) = Ax^2 + Bx + C$$

to see if it works. If so, determine the values of the constants A, B, and C needed to make it work.

1. This doesn't work for all values of x.

- **2.** Works if A = 6, B = -2, and C = 0
- **3.** Works if A = -6, B = 2, and C can be anything

- 4. Works if A = 3, B = -2, and C = 0
- 5. Works if A = -3, B = 2, and C can be anything
- 6. Works if  $A = \pi$ ,  $B = \sqrt{17}$ , and  $C = \sqrt{-1}$ .

### Simple Harmonic Motion

Schrödinger Equation for U = 0

Equation: 
$$\frac{d^2 y(t)}{dt^2} = -\frac{k_{sp}}{m} y(t) \qquad \qquad \frac{d^2 \psi(x)}{dx^2} = -\frac{2mE}{\hbar^2} \psi(x)$$

Solution:

$$y(t) = A \sin(\omega t + \phi_0)$$
  $\psi(x) = A \sin(kx + \phi_0)$   
where  $\omega = \sqrt{\frac{k_{sp}}{m}}$  where  $k = \sqrt{\frac{2mE}{\hbar^2}}$ 

Check if Schrödinger's Equation matches de Broglie relation for U = 0:

$$E = K + \mathcal{U}^{\bullet} = \frac{1}{2}mv^2 = \frac{p^2}{2m} \qquad \Rightarrow \qquad p = \sqrt{2mE}.$$
  
de Broglie:  $\lambda = \frac{h}{p} \qquad \Rightarrow \qquad k = \frac{2\pi}{\lambda} = \frac{2\pi p}{h} = \frac{p}{\hbar} = \sqrt{\frac{2mE}{\hbar^2}} \quad \checkmark$