$Homework_3_soln$

February 12, 2025

1 PHYS 310 HW 3 Solutions

[1]: import numpy as np import matplotlib.pyplot as plt from scipy import stats

1.1 Calculus Approach

H & H : Table 4.2

$$Z(A,B) = k \frac{A^n}{B^m}$$

We will find the partial derivatives due to the two vairables, and combine in quadrature.

For A:

$$\begin{split} \alpha_{Z,A} &= \alpha_A \frac{\partial Z}{\partial A} \\ \frac{\partial Z}{\partial A} &= nk \frac{A^{n-1}}{B^m} \\ \alpha_{Z,A} &= \alpha_A nk \frac{A^{n-1}}{B^m} \\ \alpha_{Z,B} &= \alpha_B \frac{\partial Z}{\partial B} \\ \frac{\partial Z}{\partial B} &= -mkA^nB^{-m-1} \\ \alpha_{Z,B} &= -\alpha_B mk \frac{A^n}{B^{m+1}} \end{split}$$

For B:

Combine (squaring the negative number doesn't matter, and we can take out the equation Z)

$$\begin{split} \alpha_z &= \sqrt{\left(\alpha_A n k \frac{A^{n-1}}{B^m}\right)^2 + \left(-\alpha_B m k \frac{A^n}{B^{m+1}}\right)^2} \\ \alpha_z &= k \frac{A^n}{B^m} \sqrt{\left(\alpha_A n \frac{1}{A}\right)^2 + \left(\alpha_B m \frac{1}{B}\right)^2} \\ \frac{\alpha_z}{Z} &= \sqrt{\left(n \frac{\alpha_A}{A}\right)^2 + \left(m \frac{\alpha_B}{B}\right)^2} \end{split}$$

2 Functional Approach to Pendulum

When does it matter if we use the functional approach with $T + \Delta T$ or $T - \Delta T$?

```
[2]: def g(L,T):
    return 4*np.pi**2 *L/T**2
L = 0.96
T = 1.970
DeltaT = 0.004
print('The best value of g:',g(L,T))
print('Ag(L,T+AT):',g(L,T+DeltaT) - g(L,T))
print('Ag(L,T-AT):',g(L,T-DeltaT) - g(L,T))
```

```
The best value of g: 9.765590687774262
Δg(L,T+ΔT): -0.03953676381878424
Δg(L,T-ΔT): 0.03977833230749894
```

It doesn't matter to one significant digit.

```
[3]: DeltaT2 = 0.15
```

```
print(\Delta g(L,T+\Delta T):,g(L,T+DeltaT2) - g(L,T))
print(\Delta g(L,T-\Delta T):,g(L,T-DeltaT2) - g(L,T))
```

```
Δg(L,T+ΔT): -1.3330344177085962
Δg(L,T-ΔT): 1.6760470673830667
```

If the uncertainty is much bigger, then the first significant digit would be different in each case.

We can be a little more quantitative about this by considering higher order terms in our Taylor series expansions of g(L,T):

$$\begin{split} g(L,T+\Delta T) &\simeq g(L,T) + \frac{\partial g}{\partial T} \,\Delta T + \frac{1}{2} \frac{\partial^2 g}{\partial T^2} \Delta T^2 \\ g(L,T-\Delta T) &\simeq g(L,T) - \frac{\partial g}{\partial T} \,\Delta T + \frac{1}{2} \frac{\partial^2 g}{\partial T^2} \,\Delta T^2. \end{split}$$

These expressions will have the same magnitude when the terms quadratic in ΔT are small compared to the linear terms, i.e.,

$$\frac{1}{2}\frac{\partial^2 g}{\partial T^2}\Delta T^2 \ll \frac{\partial g}{\partial T}\,\Delta T.$$

Rearranging this gives the condition on ΔT :

$$\Delta T \ll 2 \frac{\frac{\partial g}{\partial T}}{\frac{\partial^2 g}{\partial T^2}}$$

In this problem we have

$$\frac{\partial g}{\partial T} = -\frac{8\pi^2 L}{T^3} \quad \text{and} \quad \frac{\partial^2 g}{\partial T^2} = \frac{24\pi^2 L}{T^4}$$

so our condition on the size of ΔT becomes

$$\Delta T \ll \frac{2}{3}T.$$

This means that in this problem the two uncertainties will be the same when $\Delta T \ll \frac{2}{3}T$, or when ΔT is, say, less than a tenth of $\frac{2}{3}T \sim 0.13$. Guesses around 0.1 are close.

2.1 Functional Approach to van der Waals

H & H : Example 4.2.2

```
[4]: def P(Vm,T):
    '''Returns pressure a function of measured volume and temperature'''
    return R*T/(Vm - b) - a/Vm**2
```

```
[5]: R = 8.3145
Tbest = 298.0
alpha_T = 0.2
Vmbest = 2.000e-4
alpha_V = 0.003e-4
a = 1.408e-1
b = 3.913e-5
```

[6]: Pbest = P(Vmbest, Tbest)

```
#we can turn all the uncertainties due to each parameter into a vector u
u = np.zeros(2)
u[0] = P(Vmbest + alpha_V, Tbest)
u[1] = P(Vmbest, Tbest + alpha_T)
u = u - Pbest
print('P [MPa]:', Pbest/1e6,'/n')
print('uncertainties in P due to uncertainties in V and T:',u/1e6,'\n')
print('fractional uncertainties in P due to uncertainties in V and T:',u/1e6,'\n')
unc = np.sqrt(np.sum(u**2))
print('total uncertainty [MPa]:', unc/1.e6)
```

P [MPa]: 11.882007832411261 /n uncertainties in P due to uncertainties in V and T: [-0.01813283 0.01033692]

fractional uncertainties in P due to uncertainties in V and T: [-0.00152607 0.00086996]

total uncertainty [MPa]: 0.020872267575327286

Thus: $\alpha_P^T \sim 0.010$ MPa, $\alpha_P^V \sim 0.018$ MPa, $P(\overline{V}_{in}, \overline{T}) \pm \alpha_P = 11.88 \pm 0.02$ MPa As expected!

2.2 Calculus Approximation to van der Waals

$$\begin{split} P(V_m,T) &= \frac{RT}{V_m-b} - \frac{a}{V_m^2} \\ \alpha_P &= \sqrt{\left(\alpha_P^T\right)^2 + \left(\alpha_P^V\right)^2} \end{split}$$

$$\alpha_P^T = \left(\frac{\partial P}{\partial T}\right)\alpha_T = \frac{R}{V_m - b}\alpha_T$$

$$\begin{split} \alpha_P^V &= \left(\frac{\partial P}{\partial V}\right) \alpha_V = \left[RT\left(\frac{-1}{(V_m - b)^2}\right) - \frac{a(-2)}{V_m^3}\right] \alpha_V = \left[\frac{-RT}{(V_m - b)^2} + \frac{2a}{V_m^3}\right] \alpha_V \\ \alpha_P &= \sqrt{\left(\frac{R}{V_m - b}\alpha_T\right)^2 + \left(\left[\frac{-RT}{(V_m - b)^2} + \frac{2a}{V_m^3}\right]\alpha_V\right)^2} \end{split}$$

```
[7]: # Using the information stored above
     print('P [MPa]:', P(Vmbest,Tbest)/1e6,'/n')
     alphaPV = (-R*Tbest/((Vmbest - b)**2) + 2*a/Vmbest**3)*alpha_V
     alphaPT = (R/(Vmbest - b))*alpha_T
     print('uncertainties in P due to uncertainties in V and T:',alphaPV/1e6,alphaPT/
      ⇒1e6, '\n')
     unc = np.sqrt(alphaPV**2+alphaPT**2)
     print('total uncertainty [MPa]:', unc/1.e6)
    P [MPa]: 11.882007832411261 /n
    uncertainties in P due to uncertainties in V and T: -0.018162585626427412
    0.010336918008329707
    total uncertainty [MPa]: 0.020898119306488768
    Thus:
    \alpha_P^T \sim 0.010 MPa,
    \alpha_P^V \sim 0.018 MPa,
    P(\overline{V}_{in},\overline{T}) \pm \alpha_P = 11.88 \pm 0.02 \text{ MPa}
    As expected!
```

2.3 Monte Carlo Simulation of van der Waals

```
[8]: numExp = 1000  # Number of Monte Carlo "experiments"
Tvalues = stats.norm.rvs(Tbest, alpha_T, size=numExp)
Vvalues = stats.norm.rvs(Vmbest, alpha_V, size=numExp)
simData = P(Vvalues,Tvalues) # simulate a bunch of pressures
Pbest = np.mean(simData)
alphaP = np.std(simData, ddof = 1)
Pbest/1e6, alphaP/1e6
```

[8]: (np.float64(11.881505136113706), np.float64(0.020050128301167494))

Thus: $P(\overline{V}_{in}, \overline{T}) \pm \alpha_P = 11.88 \pm 0.02 \text{ MPa}$ As expected! Note: We don't get information about how much T and V_{in} affect the pressure individually.

2.4 H & H : 4.8

The volume flow rate is given by

$$\frac{dV}{dt} = \frac{\pi \rho g h r^4}{8\eta L},$$

First, we need to rewrite this relation in terms that will enable us to calculate the viscosity η from the length L, the radius r and the slope of a plot of flow rate F = dV/dt versus h. Let's call that slope m to simplify the notation.

The slope of the flow rate vs. h graph dV/dt = mh is

$$m = \frac{\pi \rho g r^4}{8\eta L},$$

and solving for the viscosity gives

$$\eta = \frac{\pi \rho g r^4}{8ml}.$$

Since we know the fractional uncertainties and the equation, we are using the calculus approximation. The uncertainty in the viscosity is therefore given by

$$\begin{aligned} \alpha_{\eta}^{2} &= \left(\frac{\partial\eta}{\partial m}\right)^{2}\alpha_{m}^{2} + \left(\frac{\partial\eta}{\partial L}\right)^{2}\alpha_{L}^{2} + \left(\frac{\partial\eta}{\partial r}\right)^{2}\alpha_{r}^{2} \\ &= \left(\frac{-\eta}{m}\right)^{2}\alpha_{m}^{2} + \left(\frac{-\eta}{L}\right)^{2}\alpha_{L}^{2} + \left(\frac{4\eta}{r}\right)^{2}\alpha_{r}^{2}, \end{aligned}$$

and the fractional uncertainty is given by

```
\left(\frac{\alpha_{\eta}}{n}\right)^2 = \left(\frac{\alpha_m}{m}\right)^2 + \left(\frac{\alpha_L}{L}\right)^2 + \left(4\frac{\alpha_r}{r}\right)^2
[9]: Frac_alpha_m = 0.07 # Fractional uncertainty in slope
      Frac alpha L = 0.005 # Fractional uncertainy in L
      Frac_alpha_r = 0.08
                                     # Fractional uncertainty in r
      Frac_alpha_eta = np.sqrt(Frac_alpha_m**2+Frac_alpha_L**2+(4*Frac_alpha_r)**2)
      print(Frac_alpha_eta)
```

0.32760494501762333

So, the fractional precision to which the viscosity is known is 33%. Note that this is dominated by the error in the radius measurement. If the fractional error were zero for both the slope and the length L measurement, we'd get a fractional uncertainty of 32% with the radius measurement alone because of the factor of 4 that comes in from the fact that we take the 4th power of r in the relation.

So, clearly if we want to improve the measurement, the radius of the tube (iii) is the thing that we need to work on – that dominates the error.

$\mathbf{2.5}$ H & H : 4.10

To combine the students' data best, we need a weighted mean.

$$w_i = \frac{1}{\alpha_i^2}$$
$$\overline{x} = \frac{\sum_i w_i x_i}{\sum_i x_i^2}$$
$$\frac{1}{\alpha_x^2} = \sum_i w_i$$

[10]: speeds = np.array([3.03, 2.99, 2.99, 3.00, 3.05, 2.97]) uncertainties = np.array([0.04, 0.03, 0.02, 0.05, 0.04, 0.02])weights = 1/uncertainties**2 weightedAvg = np.dot(speeds,weights)/np.sum(weights) alpha speeds = np.sqrt(1/np.sum(weights)) print(weightedAvg, alpha_speeds)

2.9921259842519685 0.011351102608219766

An average of $(2.99 \pm 0.01) \times 10^8$ m/s is pretty good. Adding (3.0 ± 0.3) is not going to be doing much because the large uncertainty causes it to be weighted lower, and it's close to the average. (4.01 ± 0.01) is going to affect it more because the claimed small uncertainty will cause it to be weighted more, and it's very far from the current average.

[11]: speeds = np.array([3.03, 2.99, 2.99, 3.00, 3.05, 2.97, 3.0]) uncertainties = np.array([0.04, 0.03, 0.02, 0.05, 0.04, 0.02, 0.3])

```
weights = 1/uncertainties**2
weightedAvg = np.dot(speeds,weights)/np.sum(weights)
```

```
alpha_speeds = np.sqrt(1/np.sum(weights))
print(weightedAvg, alpha_speeds)
```

2.992137240886347 0.011342985980361566

 $(2.99 \pm 0.01) \times 10^8$ m/s is no change.

```
[12]: speeds = np.array([3.03, 2.99, 2.99, 3.00, 3.05, 2.97, 3.0, 4.01])
      uncertainties = np.array([0.04, 0.03, 0.02, 0.05, 0.04, 0.02, 0.3, 0.01])
      weights = 1/uncertainties**2
      weightedAvg = np.dot(speeds,weights)/np.sum(weights)
```

```
alpha_speeds = np.sqrt(1/np.sum(weights))
print(weightedAvg, alpha_speeds)
```

```
3.564864020006252 0.007501172149729748
```

This really changed things $(3.565 \pm 0.008) \times 10^8$ m/s

2.6 M & M - 200 sections

We're building on last week's simulation, and extending it to 200 lab sections.

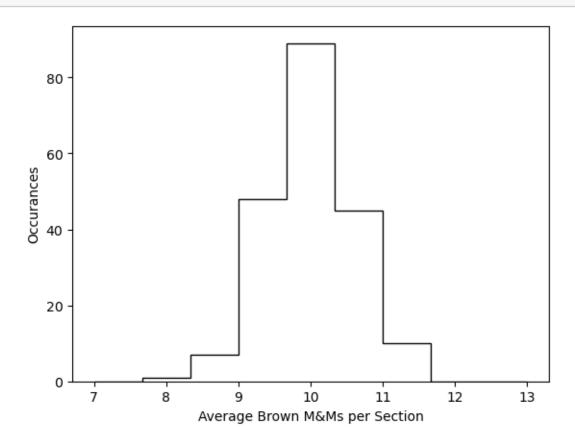
```
[13]: nmms = 60
                         # number of MEMs per bag
      nb = 24
                         # number of bags (students) per section
      nsections = 200
                         # number of sections
      #initialize arrays to write data:
      mm_means = np.zeros(nsections)
      mm_uncs = np.zeros(nsections)
      for s in range(nsections):
          #For nb bags per section, of the nmms per bag that come in six random.
       ⇔colors, count color [0] = brown M&Ms
          data_section = np.array([np.bincount(stats.randint.rvs(0,6,size=nmms))[0]_

→for i in range(nb)])

          mm_means[s] = np.mean(data_section)
          mm_uncs[s] = np.std(data_section,ddof = 1)/np.sqrt(nb)
```

[14]: binned = np.linspace(7,13,10) #define the bin edges to make it look nice

```
plt.hist(mm_means,bins = binned,histtype = 'step', color = 'black')
plt.xlabel('Average Brown M&Ms per Section')
plt.ylabel('Occurances')
```



Check the Central Limit Theorem

The standard deviation σ of the means should be equivalend to the standard error of the mean α of one section:

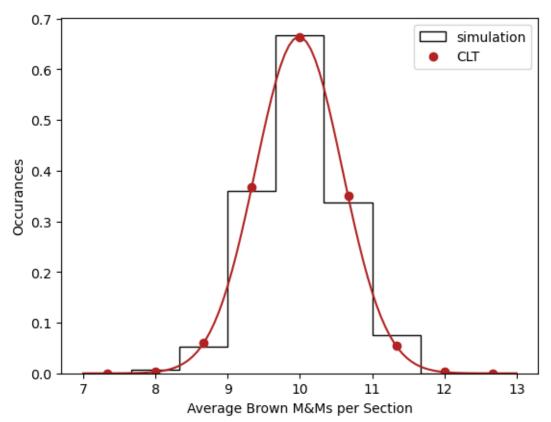
$$\sigma_{\rm averages} = \frac{\sigma}{\sqrt{N}}$$

0.6010061807496443 0.5831440892769958 The standard deviation of the means is: 0.6 brown M&Ms The typical standard error of the mean for the sections is: 0.6 brown M&Ms

The equivalence is supported.

We can also check the shape of the distribution of the means by comparing the results to the histrogram predicted by the Central Limit Theorem with a normal distribution using the mean and standard deviation.





[18]: version_information numpy, scipy, matplotlib

Software	Version
Python	3.12.8 64bit [Clang 14.0.6]
IPython	8.30.0
OS	macOS 15.3 arm 64 arm 64 bit
numpy	2.2.2
scipy	1.15.1
matplotlib	3.10.0
Wed Feb 12 18:26:27 2025 EST	
	Python IPython OS numpy scipy matplotlib

[]: